# The ornithine-based siderophore as a potential carrier of peptide nucleic acid into *Escherichia coli* cells

U. Tsylents, M. Burmistrz, P. Maj, M. Wojciechowska, J. Trylska

Centre of New Technologies, University of Warsaw, Warsaw, Poland

The rapid spread of antimicrobial resistance, especially in the case of gramnegative bacteria, encourages searching for new alternative delivery ways through the restrictive outer membrane, which is an outer layer of the bacterial envelope [1].

Iron chelators, known as **siderophores**, are secreted, recognized and later transported through the outer membrane by the TonB-dependent transport system (TBDT) [2].

**Peptide nucleic acid (PNA)** is a nucleic acid mimic with high affinity towards natural nucleic acids. PNA can be used as an antisense oligonucleotide, which binds to the target mRNA in a complementary manner regulating its expression. Gene-targeting is a promising approach against various diseases, including bacterial infections. However, due to poor cellular permeability, PNA cannot act as a gene silencing agent on its own [3].

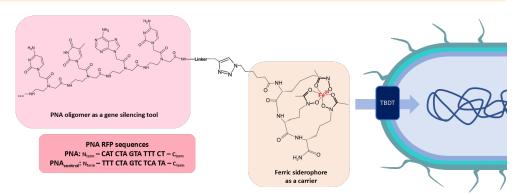


Figure 1: The Trojan horse strategy which uses the bifunctional conjugate of the synthetic siderophore and a PNA oligomer enables the passage through the *E. coli* membrane.

**Our goal** is to synthesize an ornithine-based siderophore, which upon conjugation with a PNA oligonucleotide will ensure non-invasive transport of PNA into *E. coli* cells. To confirm that, PNA anti-rfp sequence was used to target the reporter gene *mrfp1* expressing red fluorescent protein (RFP) inside *E. coli* cells (Figure 1).

Research goal

### Synthesis

Modified ornithine derivative (N<sup> $\delta$ </sup>-hydroxy-N<sup> $\delta$ </sup>-acetyl-ornithine) was used as a building block, because it provides hydroxamate groups capable of binding ferric iron.

 $S_{\rm L}$  siderophore was synthesized using the solid-phase peptide synthesis (SPPS) technique with appropriate Fmoc strategy (Figure 2).

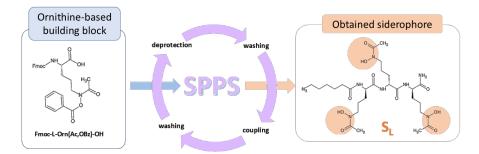


Figure 2: Chemical structures of the Fmoc-protected modified ornithine derivative used for the  $S_{\rm L}$  siderophore synthesis. Hydroxamate groups are highlighted.

In order to verify its carrier potential  $S_{\rm L}$  was later used for conjugation with PNA oligomers via copper-catalyzed azide-alkyne cycloaddition (also known as "click reaction"), which resulted in  $S_{\rm L}-{\rm PNA}$  conjugates connected via uncleavable triazole ring (Figure 1).

PNA oligomers (Figure 1) used for conjugation were synthesized manually using appropriate SPPS protocol.

# Circular Dichroism (CD) Spectroscopy

CD spectra of the synthesized siderophore mimic -  $\mathsf{S}_{\mathrm{L}}$  were recorded to verify

#### **Microbial assays**

After confirming the iron coordination properties of the S<sub>L</sub> siderophore, uptake of PNA conjugates with S<sub>L</sub> was tested on *E. coli*  $\Delta fur$  mutant with continuous iron uptake (Figure 4, left). Bacteria were cultured in iron limiting conditions. RFP fluorescence was measured using the flow cytometry with two control cultures: bacteria carrying plasmids expressing RFP (with RFP) and bacteria lacking the RFP gene (without RFP).

Additionally, unconjugated  $S_{\rm L}$  was used for the growth recovery assays. Several *E. coli* mutants lacking various TBDT receptors, proteins or reductases were used in order to investigate the  $S_{\rm L}$  pathway through outer membrane receptors.

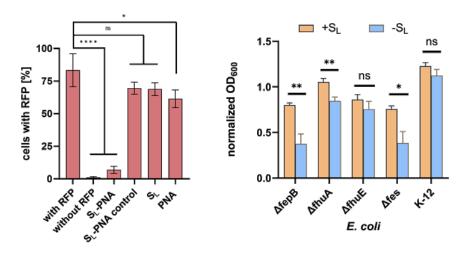
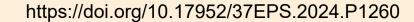


Figure 4: RFP fluorescence silencing in the *E. coli*  $\Delta fur$  mutant cultured in iron limiting conditions (left). Growth recovery assay with various *E. coli* mutants with or without 16  $\mu$ M S<sub>L</sub> in iron limiting conditions (right). Statistical significance was determined by the two-way ANOVA test (\*\*\*\*: p<0.0001, \*: p<0.05, ns: non-significant)



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iron(III)-coordination properties and later were compared to CD spectra of two natural siderophores - coprogen and ferrichrome (Figure 3). Upon binding iron(III) siderophores obtain structure, which can be assigned to  $\Delta$  or  $\Lambda$  optical isomer structure using CD analysis. Ferrichrome complex represents  $\Lambda$  configuration and coprogen obtains structure for  $\Delta$  isomer. [4]

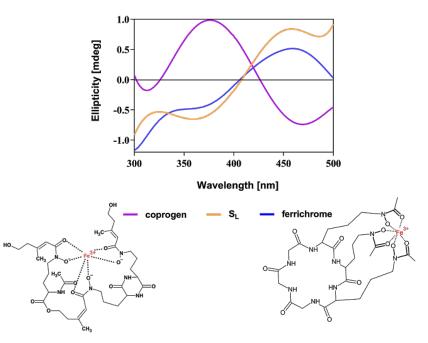


Figure 3: CD spectra of the natural and  $S_{\rm L}$  siderophores in the presence of Fe $^{3+}$  salt. Ferric complex structure of coprogen (left) and ferrichrome (right) are included. CD spectra of a natural cyclic siderophore – ferrichrome (and its ferric complex structure included on the right) is detected as  $\Lambda$  optical isomer (blue line).

Based on the CD spectra obtained for  $S_{\rm L}$ , this siderophore chelates ferric iron adopting the structure of the  $\,$  isomer (Figure 3). Similar positions of the minima and maxima in the Fe(III)-S\_L complex and ferrichrome spectra suggest that, while capturing iron,  $S_{\rm L}$  adopts a structure comparable to the cyclic ferrichrome.

#### **Conclusions**

- As shown by CD analysis,  $S_L$  efficiently binds ferric iron achieving the structure ( $\Lambda$  configuration) similar to natural cyclic siderophore ferrichrome (Figure 3, orange and purple lines).
- The growth recovery assays on various *E. coli* mutants confirms the uptake of the linear siderophore mimic via the *E. coli* receptors recognizing the hydroxamate-type siderophores.
- Conjugates with  $S_{\rm L}$  siderophore are recognized by *E. coli* iron uptake system and allow to transport PNA inside the bacterial cell.

## Acknowledgements

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#### References

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Full study here: