Peptide-based Coacervates as Evolutionary Roads of Catalytic Function

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Methodology

Introduction

Scientists suggest that cooperative interactions among peptides facilitated the formation of enzymes, introducing functionality through sequence diversification¹. While peptides serve as effective chiral catalysts in organic reactions, their utilization in aqueous reactions is still limited and remains a challenge in molecular engineering, arising from their conformational heterogeneity^{2,3}. Coacervates, formed through liquid-liquid phase separation (LLPS), are considered primitive models of protocells, linked to the origin of shielding and concentrating oligomers³. Therefore, life, the compartmentalization of catalytic peptides appears to have great potential as a route to evolve catalytic function in small catalytically modest peptides, by constraining their conformational flexibility observed in aqueous solutions.

- Coacervation was achieved by mixing 5 mg/mL of P7 peptide⁴ in 100mM of sodium phosphate buffer, containing 1M NaCl, at pH 8.0, to final peptide concentration of 1mg/mL
- This is followed by 1h of incubation at 27°C, leading to the formation of a milky-like state
- Lastly microscopy imaging is proceeded







Catalytic Microreactors



Catalyst	k _{cat} (S⁻¹)	К _м (М)	K _{cat} /K _M (M ⁻¹ S ⁻¹)
P7-based Peptide Coacervates	(4.9 ± 0.6) x10 ⁻³	(8.2 ± 3.2) x 10 ⁻⁴	5.9 ± 0.2
P7 Peptide in solution	(1.0 ± 0.0) x10 ⁻⁵	(1.4 ± 3.2) x 10 ⁻²	(4.0 ± 0.3) x 10 ⁻⁴

- 15,000-fold increase when P7 peptide coacervates are formed
- The catalytic reaction takes place within P7 coacervates

References





Acknowledgements



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- Uponcoacervateformation,P7'saffinityforphosphorylatedassembliesfacilitatesthesequestrationofphosphorylatedsupramolecules(BSAp).
- P7 coacervates demonstrate a faster sequestration of phosphorylated BSA*p* compared to non-phosphorylated BSA.