



THE UNIVERSITY OF
MELBOURNE

Tetrahydropyranyl and Tetrahydrofuranyl as New Backbone Protecting Groups For Enhanced Fmoc SPPS

Samuel J. Paravizzini, Craig A. Hutton*, John A. Karas*

Introduction

9-Fluorenylmethoxycarbonyl (Fmoc) solid-phase peptide synthesis (SPPS) is the most reliable method for preparing a wide range of peptide targets. However, synthetic efficiency often declines as chain elongation proceeds, making it difficult to routinely obtain pure peptides with >40 residues. This loss of efficiency is caused by poor chain solubility and aggregation via β -sheet formation (Figure 1a).¹

Acid-labile amide backbone protecting groups (PGs), such as the 2,4-dimethoxybenzyl (Dmb)² group, improve peptide chain solubility and disrupt hydrogen bonding networks (Figure 1b), which can significantly improve crude purity. However, benzyl-based protecting groups are often difficult to remove and the benzyl cations that form during trifluoroacetic acid (TFA) cleavages can alkylate sensitive side chains; this limits their use because multiple protecting groups are needed for the synthesis of longer sequences. Pseudoproline³ and *N,O*-benzylidene⁴ dipeptides avoid this deprotection issue, although they are largely limited to serine and threonine containing peptides.

Ideally, backbone protecting groups should be highly acid-labile and easily incorporated into a broad range of amino acid residues. Therefore, we have evaluated the use of tetrahydropyranyl (THP)⁵ and tetrahydrofuranyl (THF) groups as backbone protection for Fmoc SPPS, introduced via prepared dipeptide building blocks.

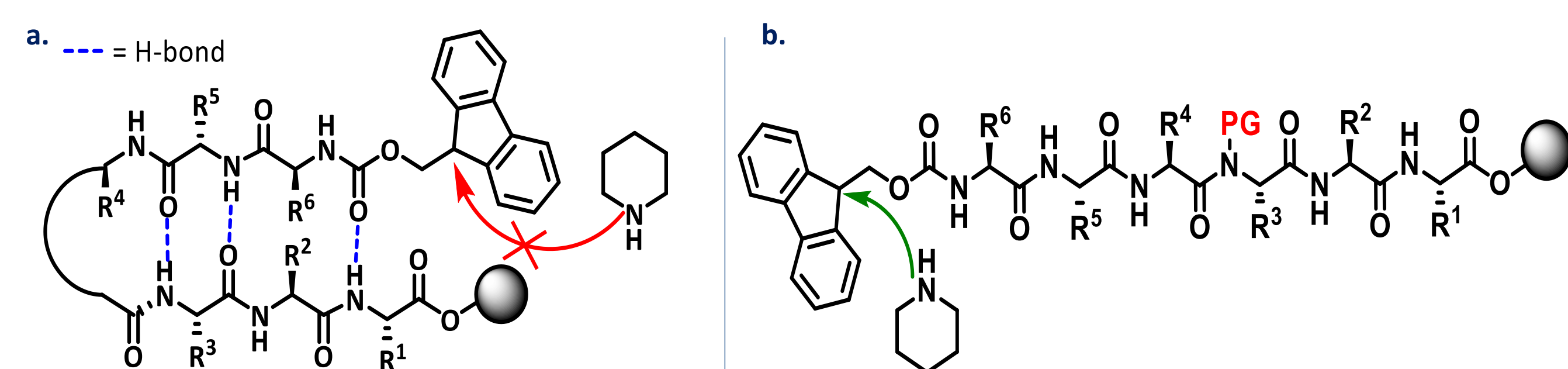


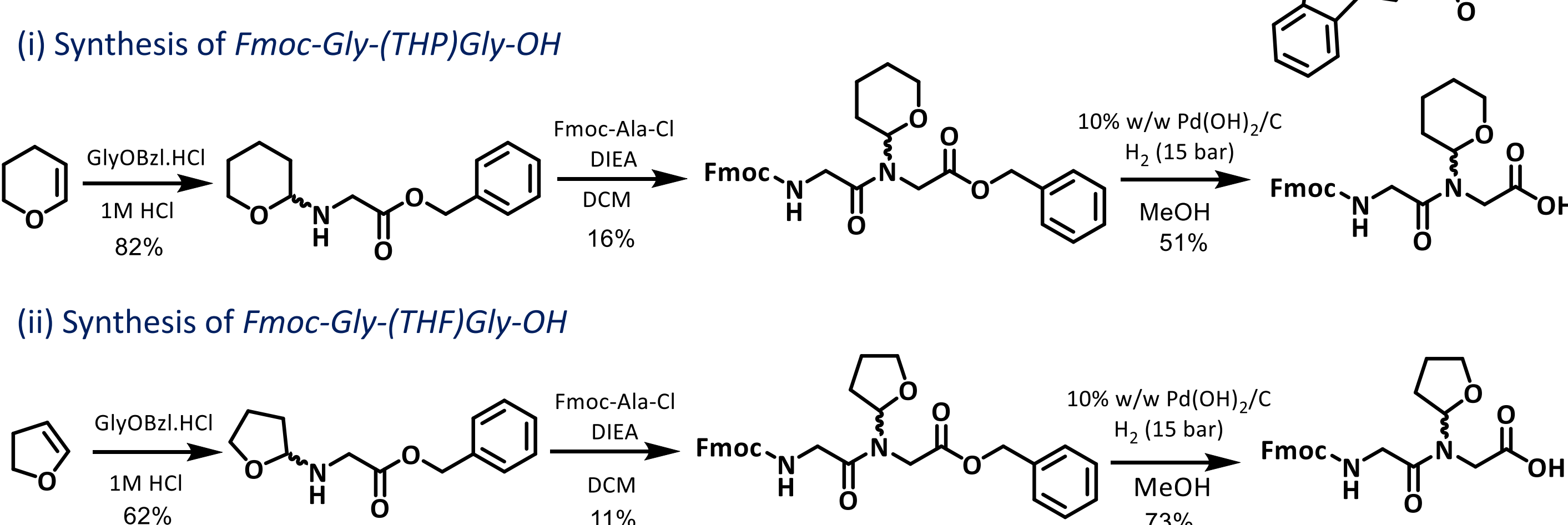
Figure 1. (a.) Backbone hydrogen bonding can cause steric crowding of the N-terminus during Fmoc SPPS. (b.) Backbone protection improves solubility and blocks hydrogen bonding, improving synthetic yields.

Objectives

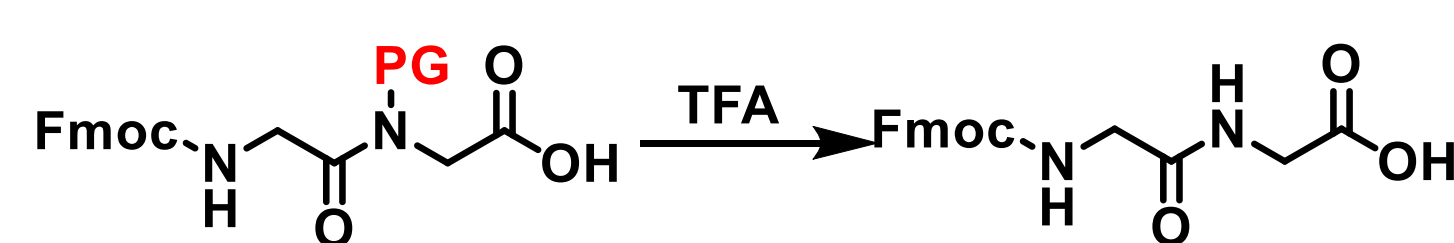
To address these limitations, we are developing new highly acid-labile backbone protecting groups that can be introduced via dipeptide precursors. Our objectives are:

- To explore the synthesis of THP and THF protected dipeptides.
- To investigate the kinetics of these protecting groups in TFA mediated cleavages.
- To evaluate their effectiveness in improving crude peptide purity during SPPS.

Synthesis of Backbone Protected Dipeptides



Acidic Cleavage Deprotection Kinetics



THP and THF protected dipeptides with general structure Fmoc-Gly-(PG)Gly-OH, were reacted in cleavage solutions containing various amounts of TFA, monitored over a 3-hour period. They were compared to the Dmb group and a pseudoproline (Ser(ψ Me₂Pro))^{*} motif.

Results:

- Tetrahydrofuran (THF) is hyper acid-labile
- Tetrahydropyran (THP) can be removed in mild cleavage cocktails
- Order of reactivity; THF > ψ Me₂Pro > THP > Dmb

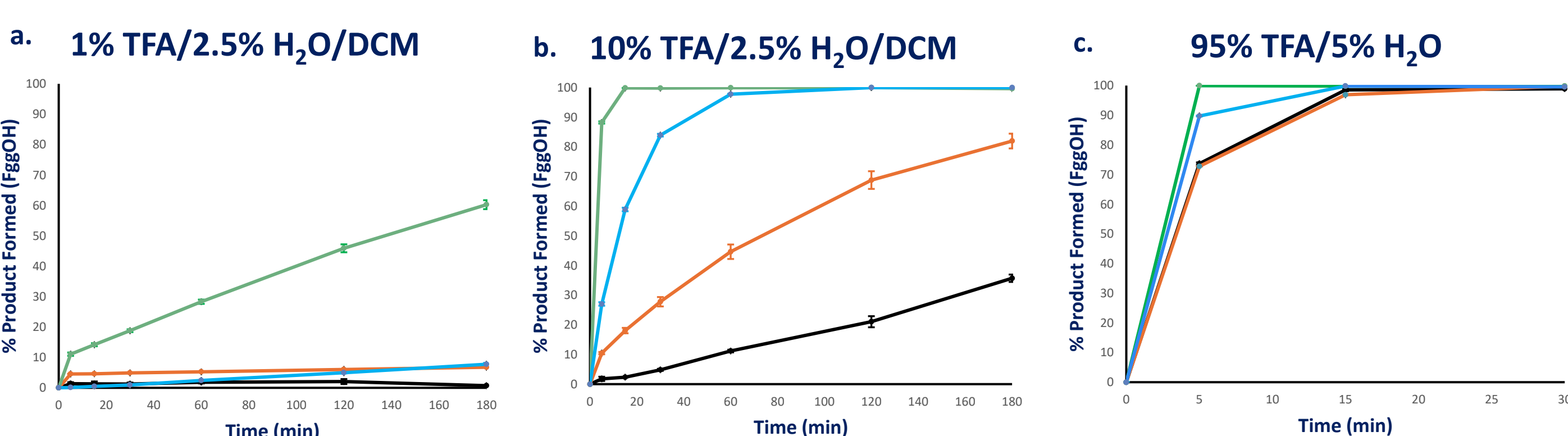
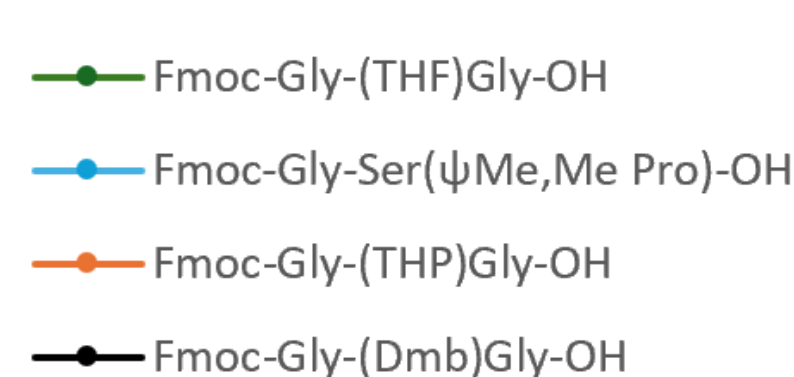


Figure 2. Deprotection kinetics of Fmoc-Gly-(PG)Gly-OH in varying amounts of TFA. (a.) 1% TFA/2.5% H₂O/DCM. (b.) 10% TFA/2.5% H₂O/DCM. (c.) 95% TFA/5% H₂O. All experiments were conducted in triplicate; error bars are calculated as standard error. * The pseudoproline was tested in the sequence Fmoc-Gly-Ser(ψ Me₂Pro)-OH.

SPPS of Amyloid A β (32-43): Fmoc-IGLMVG(PG)GVVIAT-OH

The dipeptide building blocks (Fmoc-Gly-(PG)Gly-OH) were incorporated in the SPPS of a fragment of amyloid A β protein (Figure 3.) and their improvement on the crude product purity was compared to an unprotected dipeptide control (Fmoc-Gly-Gly-OH).

Results:

- THP and Dmb provide equal improvement to crude product purity.
- THF was worse, presenting more deletion sequences closer to the site of protection.
- All the protected dipeptides provided improvement over the unprotected control.

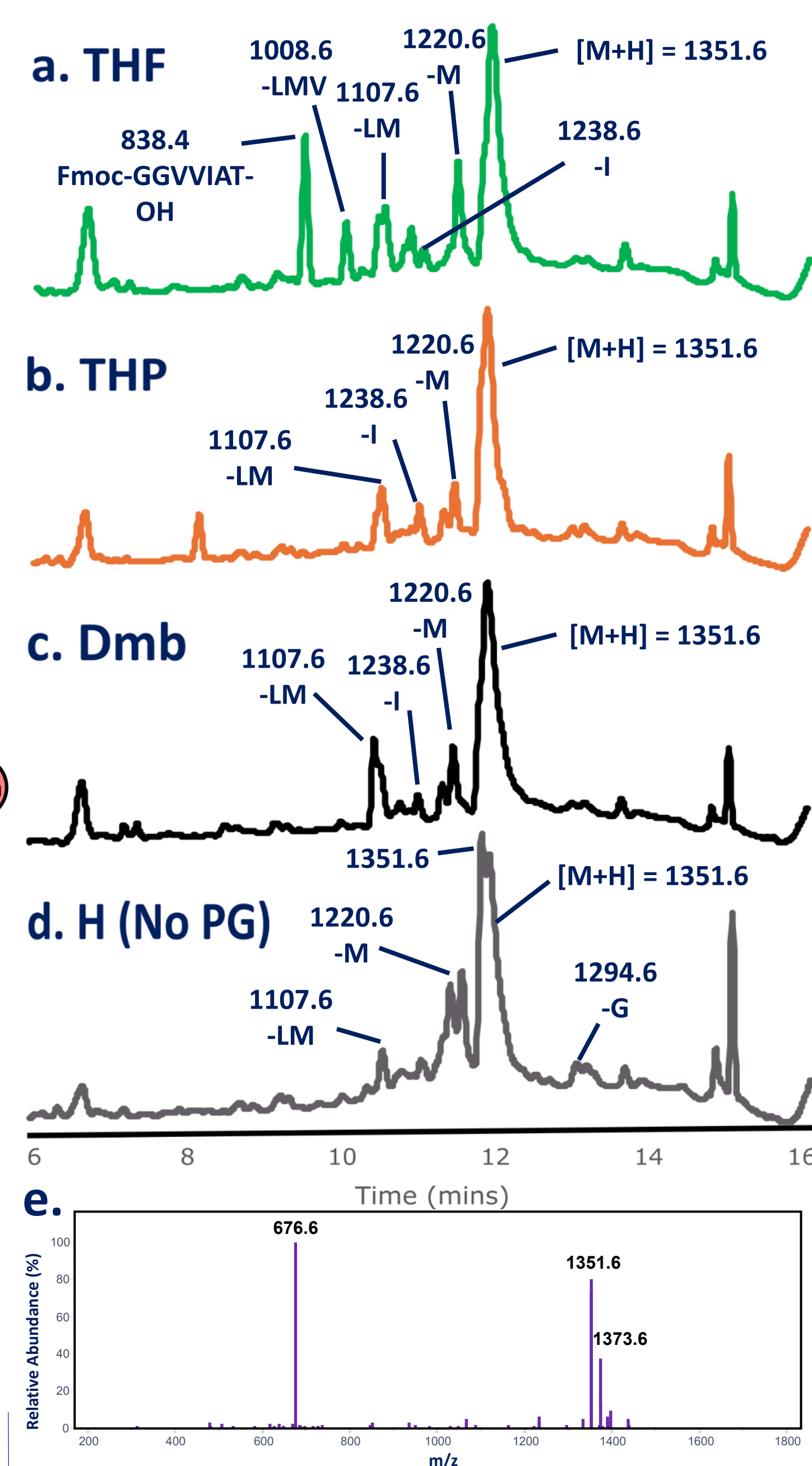
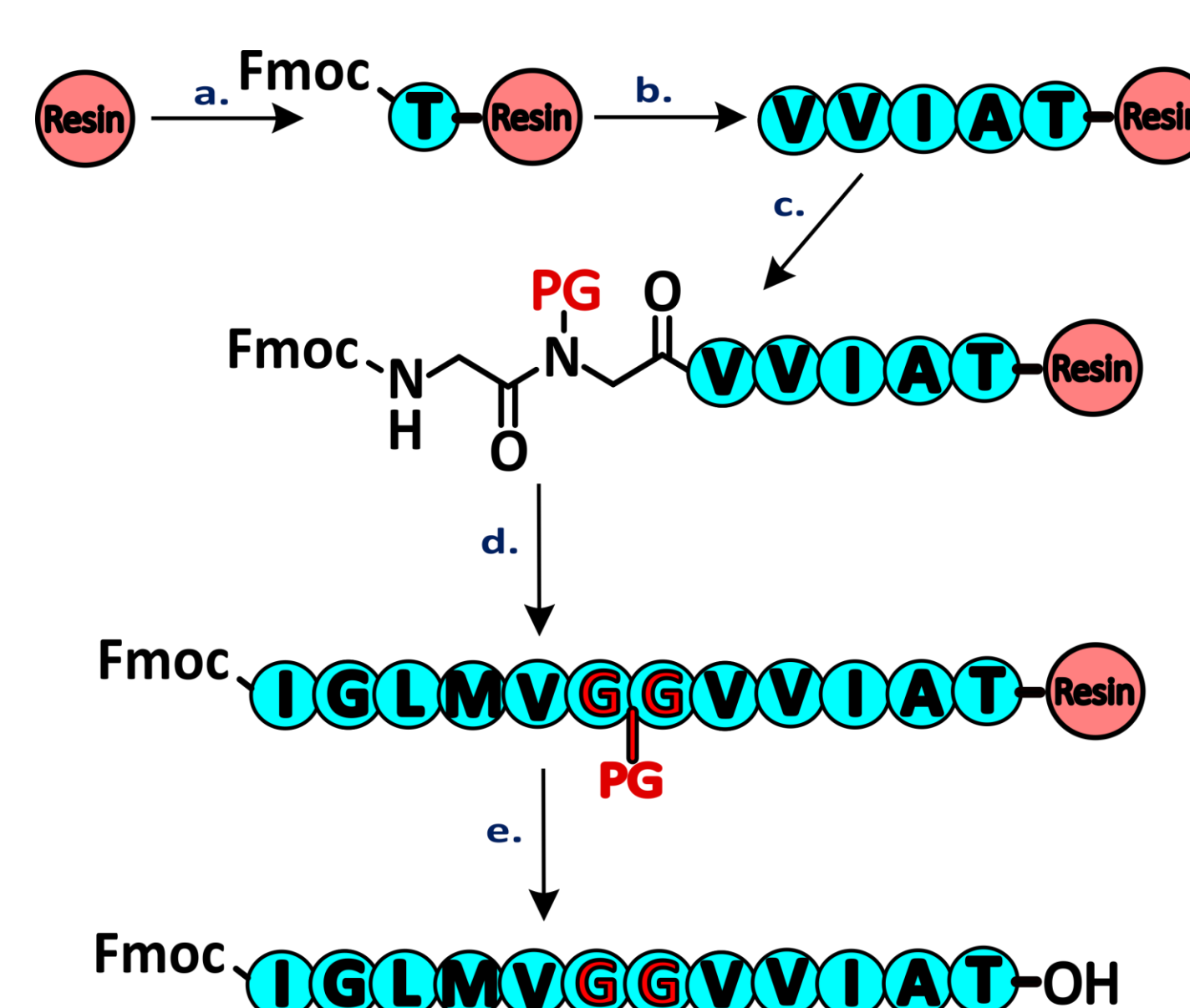


Figure 3. SPPS of amyloid A β (32-43). (a.) Loading Wang resin; 4 eq. Fmoc-Thr(OtBu)-OH, 6 eq. DIC, 0.1 eq. DMAP, DCM, 4hrs. (b.) Standard SPPS conditions; Fmoc deprotection (20% piperidine/DMF, 15 mins), then coupling (4 eq. Fmoc-Xxx-OH, 4eq. HATU, 8eq. DIEA). (c.) Protected dipeptide coupling. (2 eq. Fmoc-Gly-(PG)Gly-OH, 2eq. HATU, 4eq. DIEA). (d.) Standard SPPS conditions. (e.) Resin cleavage; 95% TFA / 2.5% H₂O / 2.5% TIPS

Figure 4. LCMS trace of Fmoc-IGLMVGGVVIAT-OH after SPPS. Dipeptides used are: (a.) Fmoc-Gly-(THF)Gly-OH. (b.) Fmoc-Gly-(THP)Gly-OH, (c.) Fmoc-Gly-(Dmb)Gly-OH, (d.) Fmoc-Gly-Gly-OH. (e.) Mass spectrum of target peak, [M+H] = 1351.6.

Conclusion

Backbone protected dipeptide building blocks are an effective way to introduce tertiary amide bonds during the SPPS of aggregation-prone sequences and provide an extended solubilising effect that greatly improves crude peptide purity. For the protecting groups investigated it was found that:

- both the THP and THF groups are exceptionally acid-labile.
- they both provide an improvement to the crude product purity during SPPS.
- the THP group can provide the same improvement to crude product purity as Dmb during SPPS.
- the THF performed worse than THP during the SPPS of A β (32-43).

Future Work

- Exploring higher yielding synthetic routes of the THP and THF protected dipeptides.
- Incorporating protecting groups into different dipeptide structures. (i.e. Fmoc-Xxx-(PG)Gly-OH or Fmoc-Xxx-(PG)Ala-OH)
- Further evaluation of side reactions occurring during SPPS. (epimerisation or diketopiperazine formation)

References

- Behrendt, R.; White, P.; Offer, J. Advances in Fmoc solid-phase peptide synthesis. *J. Pept. Sci.* 2016, 22 (1), 4-27.
- Cardona, V.; Eberle, I.; Barthélémy, S.; Beythien, J.; Doerner, B.; Schneeberger, P.; Keyte, J.; White, P. D. Application of Dmb-Dipeptides in the Fmoc SPPS of Difficult and Aspartimide-Prone Sequences. *Int. J. Pept. Res. Ther.* 2008, 14 (4), 285-292.
- Wöhr, T.; Wahl, F.; Nefzi, A.; Rohwedder, B.; Sato, T.; Sun, X.; Mutter, M. Pseudo-Prolines as a Solubilizing, Structure-Disrupting Protection Technique in Peptide Synthesis. *J. Am. Chem. Soc.* 1996, 118 (39), 9218-9227.
- Wu, H.; Sun, Z.; Li, X. N,O-Benzylidene Acetal Dipeptides (NBDs) Enable the Synthesis of Difficult Peptides via a Kinked Backbone Strategy. *Angew. Chem.* 2023, 62 (44).
- Sharma, A.; Ramos-Tomillero, I.; El-Faham, A.; Nicolas, E.; Rodriguez, H.; de la Torre, B. G.; Albericio, F. Understanding Tetrahydropyranyl as a Protecting Group in Peptide Chemistry. *ChemistryOpen*. 2017, 6 (2), 168-177.

Acknowledgements

University of Melbourne: Masson Travel Award (2023), GI Feutrill Award (2023 + 2024)
Australian Research Council

